

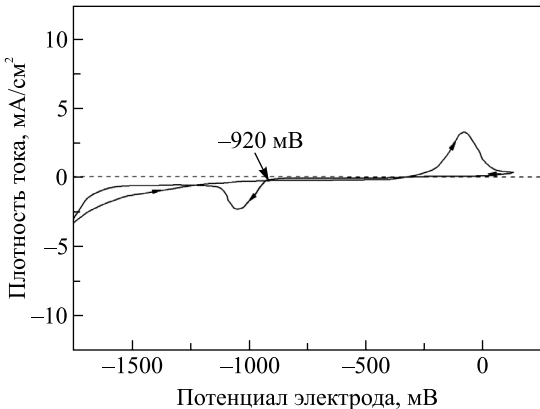


Рис 5. Вольт-амперная кривая поляризации медного электрода в растворе, моль/л:
 CuSO_4 – 0,1; NaOH – 1,5; $\text{C}_3\text{H}_8\text{O}_2$ – 0,6;
 $\text{C}_2\text{H}_8\text{O}_7\text{P}_2$ – 0,01; $\text{C}_4\text{H}_9\text{NO}$ – 0,002

Fig. 5. Volt-ampere curve of polarisation of copper electrode in solution, mol/L:
 CuSO_4 – 0.1; NaOH – 1.5; $\text{C}_3\text{H}_8\text{O}_2$ – 0.6;
 $\text{C}_2\text{H}_8\text{O}_7\text{P}_2$ – 0.01; $\text{C}_4\text{H}_9\text{NO}$ – 0.002